

## **Product** Data Sheet

# Zinc phthalocyanine

Target: Fluorescent Dye

Pathway: Others

Molecular Weight:

**Storage:** 4°C, protect from light, stored under nitrogen

\* In solvent: -80°C, 6 months; -20°C, 1 month (protect from light, stored under

nitrogen)

577.9

$$\begin{array}{c|c}
N & N \\
N & N
\end{array}$$

$$\begin{array}{c|c}
N & Zn^2 \\
N & N
\end{array}$$

### **SOLVENT & SOLUBILITY**

In Vitro

DMF: 2.7 mg/mL (4.67 mM; Need ultrasonic and warming)

| Preparing<br>Stock Solutions | Solvent Mass<br>Concentration | 1 mg      | 5 mg      | 10 mg      |
|------------------------------|-------------------------------|-----------|-----------|------------|
|                              | 1 mM                          | 1.7304 mL | 8.6520 mL | 17.3040 mL |
|                              | 5 mM                          |           |           |            |
|                              | 10 mM                         |           |           |            |

Please refer to the solubility information to select the appropriate solvent.

#### **BIOLOGICAL ACTIVITY**

Description

Zinc phthalocyanine is commonly applied in industry (catalysts, photoconductors) and biomedical (photodynamic therapy, PDT)<sup>[1]</sup>. Zinc phthalocyanine can be used to photooxidise cyclohexane<sup>[2]</sup>.

#### **REFERENCES**

[1]. Abimbola Ogunsipe, et al. Solvent effects on the photochemical and fluorescence properties of zinc phthalocyanine derivatives. Journal of Molecular Structure 650 (2003) 131–140.

[2]. NthapoSehlotho, et al. Zinc phthalocyanine photocatalyzed oxidation of cyclohexene. Journal of Molecular Catalysis A: Chemical Volume 219, Issue 2, 16 September 2004, Pages 201-207.

 $\label{lem:caution:Product} \textbf{Caution: Product has not been fully validated for medical applications. For research use only.}$ 

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